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A student heated a solid sample of $\text{Na}_2\text{CO}_3 \cdot x\text{H}_2\text{O}$ for 1 minute to remove water and determine a value for x

Figure 3 shows the apparatus used. **Table 1** shows the results recorded.

Figure 3

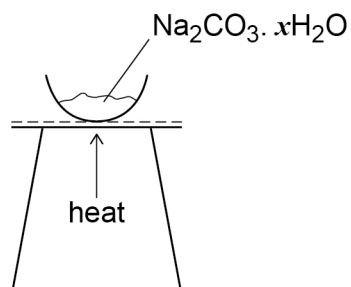


Table 1

Mass of empty evaporating basin	24.35 g
Mass of evaporating basin and solid before heating	25.47 g
Mass of evaporating basin and solid after heating for 1 minute	24.92 g

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Use the data in **Table 1** to calculate a value for x in the formula $\text{Na}_2\text{CO}_3 \cdot x\text{H}_2\text{O}$
Give your answer to 2 decimal places.

[5 marks]

Value for x _____



0 3 . 2 The correct value for x is 10

Suggest a reason for the difference between the experimental value for x and the correct value.

(If you were unable to calculate an experimental value for x assume it was 8.05. This is **not** the correct experimental value.)

[1 mark]

0 3 . 3 Suggest how the procedure could be improved, using the same apparatus, to give a more accurate value for x . Justify your answer.

[2 marks]

Suggestion _____

Justification _____

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Turn over for the next question

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