04.1 LOR Mark Scheme

Marks awarded for this answer will be determined by the quality of written communication as well as the	Indicative Chemistry Content
standard of the scientific response. Examiners should apply a 'best-fit' approach to the marking.	Stage 1 Suggested tests
Additional tests limits to lower mark within a level. This would include, for example, adding silver nitrate to the already identified sodium carbonate.	1a Add named acid to all 3
Use of hydrochloric acid with silver nitrate also limits to lower mark within a level as this would not be a logical sequence/method that would work.	1b Add water / make into a solution
Level 3 (5—6 marks)	1c Add AgNO ₃
All stages are covered and each stage is generally correct and virtually complete.	Ignore addition of NH_3 / Ignore additional test for CO_2 produced
identify all three compounds in a logical sequence with results and equations for all compounds stated.	Stage 2 Expected observations - conclusions
Covers 2 tests with matching observations, conclusions and equations	2a Na ₂ CO ₃ will fizz with acid
Level 2 (3—4 marks)	2b NaCl gives white ppt with AgNO3
All stages are covered but stage(s) may be incomplete or may contain inaccuracies OR two stages are covered and are generally correct and virtually complete.	2c NaF shows no (visible) change / no ppt
Answer is communicated mainly coherently and shows a logical progression from Stage 1 to Stages 2 and	Additional incorrect observations loses point
Covers 2 compounds	Stage 3 Equations – state symbols must match method
Isolated tests on named compounds – max LEVEL 2	$32 \text{ N}_2 \text{ CO}_1 \pm 2 \text{H}_2 \text{N}_2 \text{ N}_2 \text{N}_2 + \text{CO}_2 \pm \text{H}_2 \text{O}_2$
Level 1 (1—2 marks)	$\begin{array}{c} \text{Sa Na}_2 \text{CO}_3 + 2\text{INO}_3 \rightarrow 2\text{NaNO}_3 + \text{CO}_2 + \text{H}_2 \text{O} \\ \text{ or ionic} \end{array}$
Two stages are covered but stage(s) may be incomplete or may contain inaccuracies OR only one stage is covered but is generally correct and virtually complete.	3b AgNO ₃ + NaCl \rightarrow AgCl + NaNO ₃ or ionic
Answer includes isolated statements but these are not presented in a logical order.	3c correct state symbols